

### 3.- ISOTOPOS

Niveles de ionización magnetica del neutrino:

<http://www.reciprocalsystem.com/bpm/bpm24.htm>

<http://www.reciprocalsystem.com/ce/iratio.htm>

Los isotopos no tienen la misma abundancia en el Universo, siendo  $I$  = Nivel de ionización magnetica del neutrino, entonces el nivel de cargas isotopicas  $G$ , es:

$$G = m_v = Im_r^2/I_R = Im_r^2/156.444$$

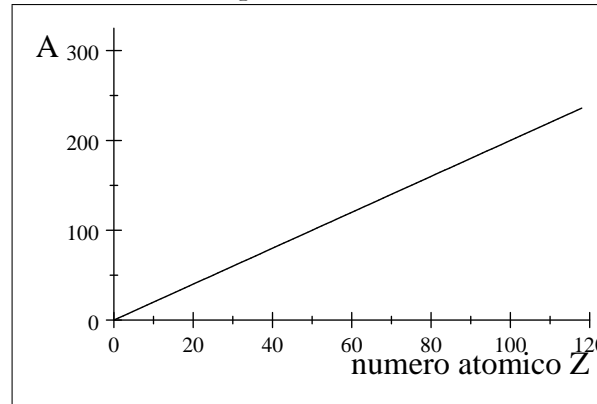
siendo interregional ratio

$$I_R = 156.444 = (1 + 2/9)128 = (1 + 2/9)2^7 = (1 + 2/9)2^3 2^2 2^2$$

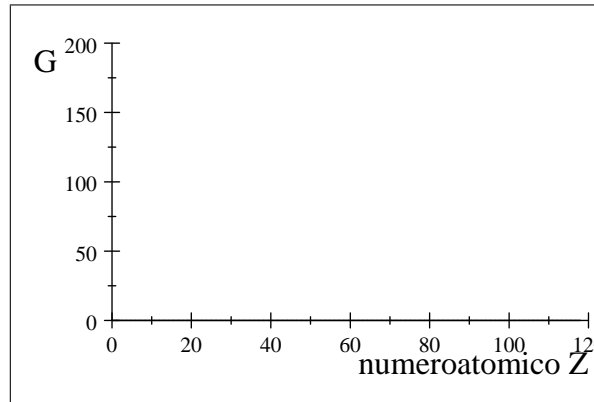
Masa vibratoria = nivel de ionización magnetica × masa rotacional<sup>2</sup> / relación interregional

$$A = \frac{\sum \text{porcentaje abundancia relativa} \times \text{masa isotopica}}{100} = 2Z + G$$

$I=0$  Para clousters globulares  $A = 2Z$

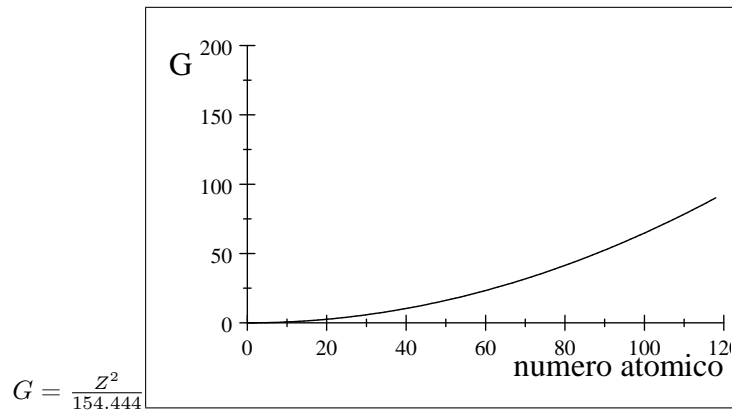


$G = 0$

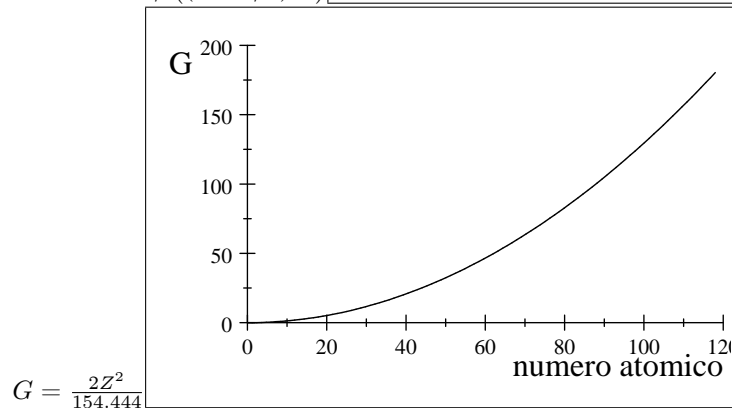
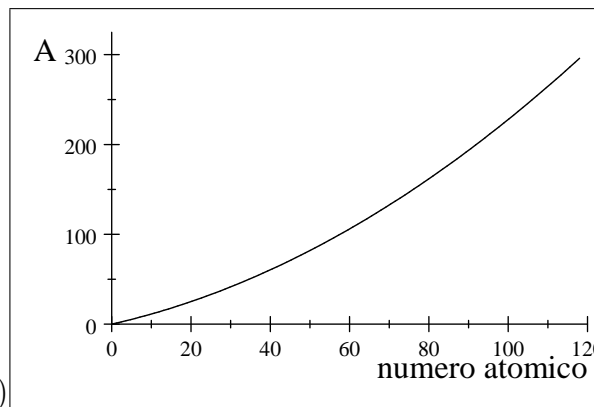


$I=1$  Para galaxia espirales

$$A = 2Z + Z^2/(1 + 2/9)2^7$$



I=2 Para galaxias esfericas



Pequeño error de Ronald W. Satz Ph. D. en la nomenclatura de los ejes de coordenadas:

Numeros magicos

$$N = 2m(m^2 + 5)/6$$

m=1,2,3,4,5,6,7,8...

Nuevo numero magico Calcio 52

<http://hyperphysics.phy-astr.gsu.edu/hbasees/nuclear/shell.html>

### Appendix 1: Neutrino Magnetic Ionization Level

Isotopes are *not* at the same abundance throughout the universe. Let  $I$  = neutrino magnetic ionization level. Then:

$I := 0$  for newly formed globular clusters in intergalactic space; all atoms have mass =  $2 \times Z$  *only*

$I := 1$  for spiral galaxies like the Milky Way at our current epoch; atoms have mass =  $2 \times Z + G$

$I := 2$  for large, end-of-life spheroidal galaxies; atoms have mass =  $2 \times Z + G$

where  $G$  = rotational vibrational mass in units of amu. Let  $m_r$  = rotational mass in terms of atomic number (total of equivalent number of electric rotational displacements) in units of amu. Then from Ref. [2], p. 264 we have:

$$G := \frac{I \cdot m_r^2}{I_R} \quad (\text{gravitational or isotopic charges}) \quad (\text{A-1})$$

This is the *midpoint* value for isotopic stability. Any mass equalling 236 ( $2 \times 118$ ) or higher will be *radioactive*.

Here are the plots for  $I = 0, 1$ , and  $2$ :

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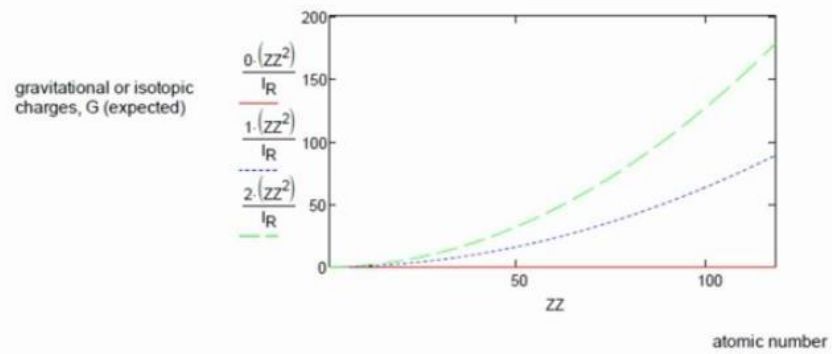


Figure 2. Isotopic Charges Expected as Function of Atomic Number



<http://www.agenciasinc.es/Noticias/Un-experimento-del-CERN-revela-un-nuevo-numero-magico-en-un-nucleo-atomico>  
<http://hyperphysics.phy-astr.gsu.edu/hbasees/nuclear/shell.html#c2>  
<http://milan.milanovic.org/math/english/atom/proton.html>  
<http://milan.milanovic.org/math/english/atom/proton.html>  
Sistema hererado  
<http://cienciaxxi.es/blog/?p=5878>