

Vamos a añadir los 6 netrones que faltan para que sea  $\text{Fe}_{56}^{26}$

$$8.3344 \times 10^{-9} J = 26 \cdot (1.5033 \times 10^{-10} J) + 32 \cdot (1.5054 \times 10^{-10} J) - 9.9466 \times 10^{-11} J \left(1 + \frac{16n^2}{26}\right), \text{ Solution is : } \{n = -2.1841\}, \{n = 2.1841\}$$

Justo n=2